

Practicing names and formulas:

Ion Combination	I or C	Formula	Name	Molar Mass
$\text{Ca}^{+2} / \text{NO}_2^-$				
$\text{B}^{+3} / \text{NO}_4^-$				
$\text{Si}^{+4} / \text{Cl}^-$				
		NH_4Cl		
		Al_2S_3		
		FeSO_4		
		SF_6		
		PF_6		
		Cl_2O_7		
		PO_5		
		H_3PO_5		
			Calcium acetate	
			Dinitrogen pentoxide	
			Manganese (II) hydroxide	
			Chromium (III) oxide	
			Nitrogen tribromide	
			Magnesium arsenite	
		HCl		
			Carbonic acid	
		CH_3COOH		
			Phosphoric acid	
		H_2S		
		H_2SO_4		
			Boric acid	
		HNO_2		