

How can the behaviour of matter be explained by the kinetic molecular theory and atomic theory?

Thoughts

- Yes The atomic theory is the theory of atoms.
- Yes The kinetic molecular theory has something to do with gases
- No We will be doing chemistry.
- Yes We will be coding a game.

There are four types of matter

- Yes Solid
- I think so Plasma
- Yes Liquid
- Yes Gas

Examples

- Yes Atoms make up the universe.
- Atoms are everywhere
- Particles are constantly in motion
- Energy makes particles move faster

There are three subatomic particles

- Negitavly charged
 - In shellElectrons
- Positively charged
 - In nucleusProtons
- Neutrally charged
 - In nucleusNeutrons

Questions

- What is the kinetic Molecular theory? Particles are constantly in motion.
- How to atoms bond?
- What is the atomic theory? The theory about what atoms are.
- How small is an atom? 100 pico meters
- How do we use Scratch? Little blocks of code
- Who discovered the atomic theory? John Dalton
- Why do atoms move fast when they heat up? Because of the kenetic enyrgy
- Who discovered the Molecular theory?
- What is plasma?
- Are there other forms of matter?
- How is this used outside of school? If you were somebody who coded for a living, if you were an engineer, or if you were a scientist.
- Is it possible to reach absolute zero?